METABOLISM OF 1 $\alpha$ -HYDROXYVITAMIN D  $_3$  TO 1 $\alpha$ ,25-DIHYDROXYVITAMIN D  $_3$  IN PERFUSED RAT LIVER

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Summary: The metabolism of 10d-hydroxyvitamin  $D_3$  (10d-OH- $D_3$ ) was studied in rat liver perfused with  $[^3H]-10d$ -OH- $D_3$ .  $[^3H]-10d$ -OH- $D_3$  was converted very rapidly to a more polar metabolite, which was identified as 10d,25-dihydroxyvitamin  $D_3$   $[^1M,25-(OH)_2-D_3]$  by co-chromatography with synthetic 10d,25- $(OH)_2-D_3$  as well as by gas chromatograph-mass spectrometry.  $[^3H]-10d$ ,25- $(OH)_2-D_3$  appeared in the perfusate as early as 20 min after addition of  $[^3H]-10d$ -OH- $^3D_3$ , and its level in the perfusate increased linearly for at least 120 min. These data strongly indicate that 10d-OH- $D_3$  is metabolized to 10d,25- $(OH)_2-D_3$ , which exerts biological effects on bone and intestine.

## INTRODUCTION

Recently, much attention has been focused on a synthetic analog of vitamin D, 100-hydroxyvitamin D<sub>3</sub> (100-OH-D<sub>3</sub>) (1-3). This analog was found to cause as great a biologic response as the hormonal form of vitamin D<sub>3</sub>, 100, 25-dihydroxyvitamin D<sub>3</sub> [100, 25-(OH)<sub>2</sub>-D<sub>3</sub>], in an in vivo time course study, suggesting that it might act directly on bone and intestine (4). However, Brumbaugh and Haussler demonstrated clearly that 100-OH-D<sub>3</sub> binds to the intestinal chromatin receptor two to three orders of magnitude less avidly than 100, 25-(OH)<sub>2</sub>-D<sub>3</sub> (5). In addition, Zerwekh et al. isolated a fraction thought to contain 100, 25-(OH)<sub>2</sub>-D<sub>3</sub> from intestinal chromatin of rachitic chicks after administration of 100-OH-D<sub>3</sub> (6). These findings, together with the fact that dihydrotachysterol<sub>3</sub>, which is similar in structure to 100-OH-D<sub>3</sub>, is rapidly hydroxylated at position 25 in vivo (7), suggest that 100-OH-D<sub>3</sub> must be converted to 100, 25-(OH)<sub>2</sub>-D<sub>3</sub> before exerting its biological effects on the target tissues. Final proof of this, however, required metabolic studies with radioactive 100-OH-D<sub>3</sub>.

Abbreviations used:  $10d-OH-D_3$ ,  $10d-hydroxyvitamin D_3$ ;  $10d,25-(OH)_2-D_3$ ,  $10d,25-dihydroxyvitamin D_3$ .

Very recently, tritiated  $10d-OH-D_3$  was successfully synthesized in our laboratory. This communication shows clearly that  $[^3H]-10d-OH-D_3$  is converted to  $[^3H]-10d,25-(OH)_2-D_3$  very rapidly in perfused rat liver.

## MATERIALS AND METHODS

Weanling male rats (Sprague-Dawley) were maintained for 6-7 weeks on a vitamin D deficient diet (8). Then their livers were carefully removed by the method of Miller et al.(9) and perfused with a mixture of heparinized blood (30 ml) from vitamin D deficient rats and Krebs-Ringer bicarbonate buffer, pH 7.4 (45 ml), equilibrated with 95%  $\rm O_2$ , 5%  $\rm CO_2$ . The perfusion apparatus (Metaloglass, Inc., Boston) was that devised by Miller et al.(9). The bile duct was cannulated and bile flowed continuously during the experiments. When the flow of perfusate through the liver (approximately 6 g wet weight) attained 30 ml/min, 1.08  $\mu$ g (411,000 dpm) of radiochemically pure [2- $^3$ H]-1 $\alpha$ -OH-D<sub>3</sub> (78 mCi/mmol, Tohira et al., in preparation) was added into the perfusate. Then 20, 40, 60, 90 and 120 min later, perfusion was terminated, and the perfusate and liver were each extracted by the method of Bligh and Dyer (10). The chloroform layers of the extracts were evaporated to dryness and chromatographed on a Sephadex LH-20 column (1.5 x 30 cm) using a solvent of 65% chloroform-35% hexane (11). All radioactive determinations were carried out with a Packard Tri-Carb, Model 3385 liquid scintillation spectrometer. Radioactivity in the bile was counted directly in dioxane scintillation fluid (12). The extracts of liver and perfusate, either before or after chromatography, were evaporated to dryness under a stream of air and counted in toluene scintillation fluid (11).

The material in the radioactive peak was co-chromatographed with authentic synthetic  $101,25-(OH)_2-D_3$  (Matsunaga et al., in preparation) on a Sephadex LH-20 column. The material suspected to be  $101,25-(OH)_2-D_3$  on the Sephadex LH-20 column was further co-chromatographed with synthetic  $101,25-(OH)_2-D_3$  on a high speed liquid chromatograph (30 cm x 6.4 mm,  $\mu$ -Bondapak  $C_{18}$ , Waters, Model 440, Milford) which was developed with methanol:water (9:1,  $\nu/\nu$ ). In

addition, a liver was perfused with 25  $\mu$ g of [ $^3$ H]-lq-OH-D $_3$  for 120 min, and then the material suspected to be 10(,25-(OH) $_2$ -D $_3$  on the Sephadex LH-20 column was purified on a Celite column (13). The purified metabolite was then identified with a gas chromatograph-mass spectrometer (Shimazu-LKB, Model 9000).

#### RESULTS

Table I shows the distribution of the total radioactivity between the perfusate, liver and bile at intervals during 120 min of perfusion. The total radioactivity in the perfusate gradually decreased, while those in the liver and bile increased during the experiment (Table I). However, the radioactivity in the bile was less than 5% of the total at each time examined, so further analyses of 10d-OH-D<sub>3</sub> metabolism were performed on extracts of the perfusate and liver only.

The chromatographic profiles of the extracts of perfusate and liver after

	Distribution					
and bile	at intervals o	during 120	min of	perfusion	with [JH]-	1 <b>%</b> -он-р <sub>3</sub>

Time	Perfusate	Liver	Bile	Total
min	dpm	dpm	dpm	dpm
20	294,000 (71.7 %)	114,000 (27.9 %)	1,600 (0.4 %)	409,600
40	224,000 (57.0 %)	162,000 (41.2 %)	6,900 (1.8 %)	392,900
60	218,000 (51.1 %)	198,000 (46.3 %)	11,300 (2.6 %)	427,300
90	165,000 (42.9 %)	204,000 (53.0 %)	15,600 (4.0 %)	384,600
120	151,000 (40.9 %)	199,000 (53.9 %)	19,200 (5.2 %)	369,200

<sup>1.08</sup> µg (411,000 dpm) of [<sup>3</sup>H]-ld-OH-D<sub>3</sub> was added to the perfusate. The amount of radioactivity recovered in the perfusate, liver and bile was always more than 90% of the isotope added. Figures in parentheses show amounts of radioactivity as percentages of the total recovered.

perfusion for 120 min are shown in Fig 1. [ $^3$ H]-1Q-OH-D $_3$  was found to be converted very rapidly to a more polar metabolite. This radioactive metabolite migrated in exactly the same position as synthetic  $^{1}$ Q,25-(OH) $_2$ -D $_3$  on co-chromatography of the extract of perfusate with synthetic  $^{1}$ Q,25-(OH) $_2$ -D $_3$  on a Sephadex LH-20 column (Fig 1, bottom) as well as on high speed liquid chromatography (data are not shown). Final proof that this radioactive metabolite was  $^{1}$ Q,25-(OH) $_2$ -D $_3$  was obtained by gas chromatograph-mass spectrometry [m/e 416 ( $^{+}$ M), 398, 380, 269, 251, 134 and 59].

When perfusate containing  $[^3H]-l\alpha-OH-D_3$  was circulated in the perfusion apparatus for 120 min in the absence of liver, no radioactivity appeared in

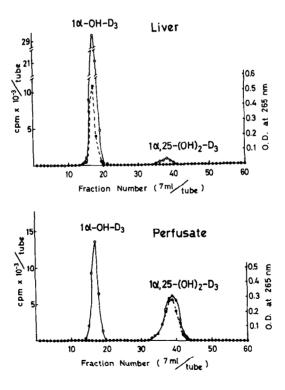


Fig 1. Chromatographic profiles on a Sephadex LH-20 column of extracts of liver (top) and perfusate (bottom) after 120 min of perfusion. Before application to the column, 200 µg of synthetic 1d-OH-D were added to the liver extract (top) and 200 µg of synthetic 1d,25-(OH) 2-D were added to the perfusate extract (bottom). The solid line shows radioactivity and the dotted line shows optical density at 265 nm.

the elution position of  $100,25-(OH)_2-D_3$ . Thus the metabolite was not produced by non-enzymatic oxidation during circulation of  $100-OH-D_3$  in the perfusate equilibrated with 95%  $O_2$ , 5%  $CO_2$ .

The amount of  $[^3H]-10I-0H-D_3$  in the perfusate decreased rapidly during the experiments with its rapid concomitant incorporation into the liver (Fig 2, left). The uptake of  $[^3H]-10I-0H-D_3$  by the liver attained a saturated level in 90 min. The radioactivity found in the liver extract was mainly unchanged  $[^3H]-10I-0H-D_3$  (Fig 1, top), and the level of  $[^3H]-10I,25-(OH)_2-D_3$  was less than 3% of the total radioactivity recovered in the perfusate, liver and bile at each time examined (Fig 2, right). On the other hand,  $[^3H]-10I,25-(OH)_2-D_3$  was found in the perfusate as early as 20 min after addition of  $[^3H]-10I-0H-D_3$ , and its level increased linearly for 120 min. At 60, 90 and 120 min after addition of  $[^3H]-10I-0H-D_3$ , 9, 14 and 19% of the total radioactivity was recovered as  $[^3H]-10I,25-(OH)_2-D_3$  in extracts of perfusate, respectively (Fig 2, right). When a liver was perfused with 1.08  $\mu$ g of  $10I-OH-D_3$  for 120 min, as much as 210 ng of  $10I,25-(OH)_2-D_3$  were recovered in extracts of the perfusate and liver. Thus the rate of  $10I,25-(OH)_2-D_3$  synthesis was approximately 17 ng/hr/g of liver.

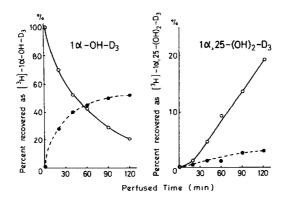


Fig 2. Time course of change in the levels of  $[^3H]-1\mathbf{d}-OH-D_3$  (left) and  $[^3H]-1\mathbf{d}',25-(OH)_2-D_3$  (right) in the total radioactivity recovered. The solid line shows radioactive metabolites in perfusate extracts and the dotted line shows those in liver extracts. Values are percentages of the total radioactivity recovered in the perfusate, liver and bile.

## DISCUSSION

The idea that  $10\text{d}-OH-D_3$  must be converted to 10d,25- $(OH)_2$ -D<sub>3</sub> before exerting its biological effects on target tissues has been suggested by several indirect lines of evidence (5-7,14). In this work we could obtain final proof of this idea using  $[^3H]-10\text{d}-OH-D_3$  synthesized in this laboratory.

At least in rats, the liver is the major site of conversion of vitamin  $D_3$  to the 25-hydroxy derivative (15,16), so we perfused the livers of vitamin  $D_3$  to the 25-hydroxy derivative (15,16), so we perfused the livers of vitamin  $D_3$  deficient rats with medium containing  $D_3$   $D_3$ 

# REFERENCES

- Holick, M. F., Semmler, E. G., Schnoes, H. K., and DeLuca, H. F. (1973) Science <u>180</u>, 190-191.
- Barton, D. H. R., Hesse, R. H., Pechet, M. M., and Rizzardo, E. (1973)
  J. Am. Chem. Soc. 95, 2748-2749.
- Kaneko, C., Yamada, S., Sugimoto, A., Eguchi, Y., Ishikawa, M., Suda, T., Suzuki, M., Kakuta, S., and Sasaki, S. (1974) Steroids 23, 75-92.
- Toffolon, E. P., Pechet, M. M., and Isselbacher, K. (1975) Pro. Nat. Acad. Sci. USA 72, 229-230.
- Brumbaugh, P. F., and Haussler, M. R. (1974) J. Biol. Chem. <u>249</u>, 1251– 1257.
- Zerwekh, J. E., Brumbaugh, P. F., Haussler, D. H., Cork, D. J., and Haussler, M. R. (1974) Biochemistry 13, 4097-4102.
- 7. Hallick, R. B., and DeLuca, H. F. (1972) J. Biol. Chem. <u>247</u>, 91-97.
- 8. Suda, T., DeLuca, H. F., and Tanaka, Y. (1970) J. Nutr. 100, 1049-1052.
- Miller, L. L., Bly, C. G., Watson, M. L., and Bale, W. F. (1951)
  J. Exptl. Med. 94, 431-453.
- Bligh, E. G., and Dyer, W. J. (1959) Can. J. Biochem. Physiol. <u>37</u>, 911-917.
- 11. Holick, M. F., and DeLuca, H. F. (1971) J. Lipid. Res. 12, 460-465.

- 12. Bray, G. A. (1960) Anal. Biochem. <u>1</u>, 279-285.
- Suda, T., DeLuca, H. F. Schnoes, H. K., Ponchon. G., Tanaka, Y., and Holick, M. F. (1970) Biochemistry 9, 2917-2922.
  Reynolds, J. J., Holick, M. F., and DeLuca, H. F. (1974) Calc. Tissue
- Res. <u>15</u>, 333-342.
- 15. Horsting, M., and DeLuca, H. F. (1969) Biochem. Biophys. Res. Commun. 36, 251-256.
- 16. Ponchon, G., Kennan, A. L., and DeLuca, H. F. (1969) J. Clin. Invest. 48, 2032-2037.